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## Microbial degradation of polyethylene terephthalate (PET): An outlook study

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### Abstract

The bacteria, known as *Ideonella sakaiensis*, is capable of producing two enzymes known as PETase and MHETase. These enzymes can degrade PET (Polyethylene terephthalate) into ethylene glycol and terephthalic acid. The genes that code for these two enzymes can be transferred to a much suitable host that has a high enough copy number that can produce the enzymes in economically viable amounts. The mass-produced enzymes can then be made to react with PET materials which will get converted to the by-products. Following this the products, ethylene glycol and terephthalic acid can then be separated. Each of the products can further be treated to convert them into energy sources that can be utilized by other microorganisms. The ethylene glycol is converted into formaldehyde which can be utilized by various methylotrophic bacteria. The terephthalic acid that is produced can be used by *Arthrobacter sp.* bacteria. Thus, helping us to achieve the complete biodegradation of PET plastic waste materials.

**Keywords:** MHETase, polyethylene terephthalate, *Ideonella sakaiensis*, PETase

### 1. Introduction

Plastic pollution is the accumulation of plastic objects and substances, in the various components of the ecosystem. It was the major threat to wildlife, habitats, and also human life associated with these ecosystems (Moore, 2020) [10]. Plastic pollutants have been classified based on their size as, micro-debris, meso-debris, and macro-debris (Hammer *et al.*, 2012) [4]. The low expense of manufacturing, high durability, and chemical structure, which makes them resistant to normal degradation procedures, have led to the rise of plastic products in our society or environment. Macro and Microplastics have shown maximum accumulation in the Northern hemisphere, mainly in waterfronts and urban centers. Plastic debris that starts as macro-debris can turn into meso or micro-debris as a result of degradation or collision (Hammer *et al.*, 2012) [4]. Plastic pollution on land poses more of a threat than what is observed in the ocean. It affects plant life and animal life, including that of the humans who are settled on the land (Anderson *et al.*, 2018) [1]. The plastic wastes that are not properly managed varies from 60% in East Asia and Pacific to 1% in North America (Hannah *et al.*, 2018) [6].

The number of plastic pollutants that are entering the ocean is increasing yearly. As of a recent survey that was carried out, it is expected that there are more amounts of plastic materials on the seabed as when compared to that of the ocean surface. The great Pacific Garbage Patch is an example of the harmful effects of plastic pollution in the ocean. According to Forbes, The Great Garbage Patch is about 1.6 million square kilometers in size as of 2020. Not just that the plastic debris that falls into the ocean destroys the aesthetic beauty of the ocean but it also causes major harm to marine life. Plastic can cause the release of toxic carcinogens such as diethylhexyl phthalate. Plastic can also pollute water bodies by the release of heavy metals such as cadmium, lead, and mercury.

Plastic pollutants not only harm the abiotic components of the ecosystems but also causes major harm to the flora and fauna of a region (aquatic or terrestrial). Plastic pollutants if accidentally consumed by organisms can likely lead to the death of the organisms as this can cause blockages within the digestive tract of the organism and ultimately lead to starvation and death of the organism. Likewise, the toxic substances that could be released as a result of plastic pollution can poison animals in a variety of ways which could in turn poison the food supply of other organisms including humans.

The plastic debris that is not visible to the naked eye could be scarier than what was anticipated earlier. Microplastics are a growing concern in our modern world. Microplastics can be of such small scales that they can be even found in the womb of pregnant women. (Ragusa *et al.*, 2021) <sup>[14]</sup>.

Plastic can also alter the chemistry of the soil drastically. The films formed as a result of plastic pollution could lead to increased evaporation of the soil of a region leading to the faster drying out of the soil of that specific area. The microplastics present in the soil could also lead to nutrient immobilization, leading to lower availability of nutrients to plants (Salt, 2019) <sup>[15]</sup>.

### 1.1 About the problem

There are many issues related to the different ways in which plastic pollutants are dealt with. Conventional methods of plastic waste disposal are methods such as landfills and incineration. As mentioned, numerous drawbacks are associated with these methods, and they can end up causing more harm than good.

The landfill is one of the oldest methods of waste disposal. When rainfall occurs in areas of landfills, the water from the rain percolates through the garbage that is present and accumulates various toxic substances as it passes down. This forms what is known as “Leachate”. If left unchecked, this leachate could cause major harm by polluting the groundwater sources. Depending on the constituents of plastic products the time it takes to degrade a plastic waste in a landfill can vary anywhere from ten years to more than centuries. Very little of the plastic waste that is being discarded daily is being recycled. Most of the plastic wastes either end up in large landfills or get discarded carelessly into open spaces. Another issue associated with discarding plastic wastes in the open is that there is a high chance of accidental consumption of the said plastic wastes by animals, which could prove to be fatal for the animal. The plastic waste that is consumed by the organism gets retained in its digestive tract and leads to the starvation of the organism and causes the death of the organism. Landfills could cause eutrophication of water bodies. It could also negatively impact the flora and fauna of a region by releasing toxic substances into the environment. Landfill leachate can affect both groundwater sources and surface water sources. The chemical composition of the leachate and the harm caused by it is mainly dependent upon various factors such as the composition of the waste, climatic conditions, age, and degradation of solid waste. The existing methods employed to treat the landfill leachate are either too expensive or can lead to the production of harmful by-products (Stefanakis *et al.*, 2014) <sup>[16]</sup>.

The leachate produced can also hinder the flourishing of soil microflora, thus reducing the quality of the soil, and adversely affecting the topography of an area. The plastic wastes that accumulate in an area can also cause the water to not percolate into the soil, which will deplete the groundwater reserve.

Another conventional way of dealing with plastic wastes is to incinerate them. There are numerous disadvantages to the burning of plastic wastes. It is deadly for both aquatic and terrestrial life. Various incineration techniques such as gasification, plasma arc, etc are employed to burn plastic wastes. Plastic at its core is petroleum-based and its incineration can cause the release of various toxic fumes which can cause various respiratory issues. The incineration of plastic can also cause the release of various greenhouse gases which can inadvertently lead to global warming. Not

only that the fumes produced as a result of plastic burning can lead to respiratory issues, but they can also lead to various skin allergies or even cancer. As a result of plastic burning, air toxicity can also rise. In many areas, this can also contribute to the phenomenon known as smog. Prolonged exposure to plastic fumes as a result of plastic burning can also have varied neurological effects such as nausea, dizziness, etc due to the release of chemical components such as styrene, Butadiene, and acetone. The melting of plastics is also not a good alternative as it can release various dioxins into the environment. Home burning of plastics, which is a common practice in many rural areas, can cause the release of various harmful gases directly into the atmosphere without any treatment or any ways to contain it. Recycling is a good way to deal with plastic wastes that are being generated. But it is to be noted that almost 90% of the plastic that is being produced is not being recycled. Also, many of the garbage patches formed in the oceans are way beyond what can be easily recycled and dealt with. Another issue that of recycling plastic wastes is that it does not essentially reduce the amount of plastic that is present in the world. It is just being transformed from one form to another. As mentioned previously the complete degradation of any plastic wastes causing its complete breakdown is by burning or incinerating it and we saw why it is not at all safe to burn plastic wastes.

### 1.2 Objectives of the study

The study aims to develop a protocol for the complete biodegradation of plastic wastes using microorganisms and their enzymes. The objectives are,

1. Successful cultivation of bacteria *Ideonella sakaiensis* on a suitable growth medium.
2. Identification and Isolation of genes that code for the enzymes PETase and MHETase from the bacteria.
3. Cloning and expression of the genes in suitable vectors.
4. Separation of enzymatic degradation products of PET degradation as a result of bacterial action.
5. Conversion of ethylene glycol to formaldehyde using periodic acid.
6. Culturing of microorganisms on formaldehyde for its complete assimilation.
7. Culturing of microorganisms on terephthalic acid for its complete assimilation.

### 2. Review of literature

*Ideonella sakaiensis* strain 201-F6 is a bacteria that belong to the genus *Ideonella*. It has displayed the ability to break down polyethylene terephthalate (PET) and use it as a source of carbon (Yoshida *et al.*, 2016 March) <sup>[20, 21]</sup>. *I. sakaiensis* was first discovered in Japan, in 2016. It was discovered by Kohei Oda and Kenji Miyamoto.

*I. sakaiensis* is a gram-negative bacterium. It is aerobic. Its morphology is rod-shaped. Cells show motility which is due to the presence of a single flagellum. *Ideonella sakaiensis* responds positively to both catalase and oxidase tests. *Ideonella sakaiensis* grows in a temperature range of 15-42 degrees celsius with an optimum of 37 degrees celsius. It grows at a pH range of 5.5-9 with an optimum of 7-7.7. Colony characteristics of *Ideonella sakaiensis* are as follows: colony colour- colourless; colony texture- smooth; colony shape- circular; colony size- 0.6-0.8 micrometre width and 1.2-1.5 micrometer length. It is observed that the bacteria can grow on PET surfaces by adhering to PET surfaces with the aid of thread-like appendages. Apart from the fact that these appendages serve the function of aiding in attachment, it also





PET surface and make use of an enzyme, PET hydrolase or PETase.

The PETase degrades PET and converts it into mono (2-hydroxyethyl) terephthalic acid (MHET). This MHET is a heterodimer that is made up of two major constituents, Ethylene glycol and terephthalic acid (TPA). The PET hydrolase or PETase found in *I. sakaiensis* functions by hydrolyzing the ester bonds that are found in PET. Upon further analysis, it was observed that *Ideonella sakaiensis* was able to produce a second enzyme called the MHETase enzyme or MHET hydrolase. This enzyme was found on the outer cell membrane. The MHETase is capable of degrading MHET, which is produced as a result of hydrolysis of PET by the PETase enzyme that was produced in the early stages of degradation by the organism, into Ethylene glycol and terephthalic acid (O. Kohei *et al.*, 2016). PETase and MHETase share a similar structure and can be combined to form a super enzyme which can increase the activity of the enzymes. This was first discovered at the University of Portsmouth, United Kingdom. The super enzyme was formed as a result of the combination of the two separate enzymes, PETase and MHETase (Carrington, 2020) [2]. Polyethylene terephthalate hydrolase or PETase is coded by the gene ISF6\_4831 and MHETase is coded by the enzyme ISF6\_0224 (UniProtKB - A0A0K8P6T7 (PETH\_IDESA) and UniProtKB - A0A0K8P8E7 (MHETH\_IDESA)). According to the above-mentioned Shotgun sequences, Gene ISF6\_4831 is observed in the 74th contig and the gene ISF6\_0224 is observed in the 104th contig. The whole-genome shotgun sequence consists of 227 contigs. The whole organism consists of 6,142,063 bp. It has a total number of 5,527 proteins in it (NCBI accession number: BBYR01000000).

## 2.1 Polyethylene terephthalate Hydrolase (PETase)

PETase belongs to the class of enzymes designated as an esterase. Esterases are enzymes that hydrolyse catalyze the splitting of esters into acid and alcohol with the aid of water. PETase is responsible for the catalysis of PET into its monomeric form mono-2-hydroxyethyl terephthalate (MHET) (Yoshida *et al.*, 2016) [20, 21]. The PETase was first observed in *Ideonella sakaiensis* strain 201-F6. It was discovered in 2016. The structure of PETase consists of three polypeptide chains, and these polypeptide chains were observed in an asymmetric unit. These three chains are denoted as A, B, and C chains. PETase adopts an alpha/beta-hydrolase fold. Within the fold is the highly conserved catalytic triad S131-H208-D177, found on the protein surface. The S131 part is what acts as the nucleophile. This nucleophile is located within a short distance that it can be polarized by the base H208. This is stabilized by acid D177. Many unique features are associated with the catalytic center of the enzyme. For instance, PETase forms two intramolecular disulphide bridges. These disulfide bridges are named as DS1 and DS2. This is only seen in PETase as all the other homologous enzymes only have one Disulfide bridge. The DS1 is what is specific to the PETase. DS2 connects the C-terminal helix and the last loop, while the DS1, which is only found in PETase, links the two loops that are associated with the catalytic acid and the base (Han. X *et al.* 2017) [5].

PETase is coded by the gene ISF6\_4831. The gene is observed in the 74th contig out of the 227 contigs. The sequence that codes for the enzyme is as follows:

```
1321 ctctgacac ctacaccaac aggagacaac atgaactttc cccgccttc cgccctgat
1381 caggccgccc ttctggcgg gctgatggcc gtgtcgccg cegccaccgc ccagaccaac
1441 ccctacgcc gcggccgaa cccgacagcc gctcactcg aagccagcgc cggcccgtc
```

```
1501 accgtgcgct cgttaccgt gagccgccc agcgctcag gcgcccggac cgtgtactac
1561 cccaccaacg ccggcggcac cgtggcgccc atcgccatcg tggcgggcta caccgcgccc
1621 cagtcgagca tcaaatgggt gggcccgcgc ctggcctcgc acggcttcgt ggtcatcacc
1681 atcgacacca actccagct cgaccagccg tccagccgct cgtcgcagca gatggccgcg
1741 ctgcgcccag tggcctcgt caacggcacc agcagcagcc cgtactacgg caaggtcgac
1801 acccccgcga tggcgtgat gggctgtctg atggcggtg gcggctcgt gatctggcgc
1861 gccacaacc cgtcgtgaa agcccgccgc ccgagccccc cgtgggacag ctcgaccaac
1921 ttctcgtgc tcaccgtgcc cagctgatc ttccctgctc agaacgacag ctcgcccgcg
1981 gtaactcgt ccgcccctgc gatctagac agcagtgcgc gcaatgcgaa cgacttctc
2041 gagatcaac gtggctgca ctctcgcgc aacagccgca acagcaacca ggcgctgat
2101 ggcaagaag gcgtggcctg gatgaagcgc ttcattgaca acgacagcgc ctactccacc
2161 ttgcctcgc agaaccgaa cagcaccgc gtgtggact tccgaccgc gaactcgacc
2221 tgagtctac cgtttcaat caggcgtgat gccttgggt gcccgccgc gcaagagccg
The gene sequence is made up of 872 base pairs. The translation of the above-mentioned nucleotide yields a result as seen below:
```

```
MNFPASRLMQAAVLGGLMAVSAATAQTNPYARGPNPTAASLE
ASAGPFTVRSFTVSRPSGYGAGTVYYPTNAGGTVAIAIVPGYTARQS
SIKWWGPRLA
SHGFVVITIDTNSLTDQSSRSSQMAALRQVAVSLNGTSSSPIYGVKVD
ARMGVMGWWS
MGGGSLISAANNPSLKAAAPQAPWDSSTNFSSVTVPTLIFACENDSI
APVNSSALPI
YDSMSRNAKQFLEINGGSHSCANSNGSNQALIGKKGVAWMKRFMD
NDRYSTFACENP
```

```
NSTRVSDFRANCS
```

There are a total of 290 amino acids in the sequence. The above-mentioned nucleotide and amino acid sequence can be obtained from NCBI using the accession ID: BBYR01000074.

The structure has been obtained from Uniprot, Accession ID: A0A0K8P6T7.

The enzyme, as mentioned previously is an esterase, more specifically it is a lipase.

## 2.2 Mono (2-hydroxyethyl) terephthalic acid hydrolase (MHETase)

MHETase belongs to the class of enzymes designated as hydrolases. It was first observed in 2016. As the name suggests the enzyme causes the hydrolysis of Mono(2-hydroxyethyl) terephthalic acid, which was formed as a result of the action of PETase enzyme on PET, into terephthalic acid and ethylene glycol (Yoshida *et al.*, 2016) [20, 21]. The enzyme was first observed in *Ideonella sakaiensis* strain 201-F6. The structure of MHETase shares a close resemblance to tannase and feruloyl esterase. Like PETase, MHETase also displays an alpha/beta-hydrolase fold. Within the fold, the catalytic triad was formed by S225, H528, and D492. The catalytic triad is flanked by the presence of 5 disulphide bonds. The disulphide bonds also flank an oxyanion which comprises backbone amide nitrogen atoms of G132 and E226. Unlike PETase, the catalytic site of MHETase binds to the substrate very tightly. Structural analysis results of MHETase identified that the lid domain of the enzyme was the major difference to the closely related tannases and feruloyl esterases. MHETase might originate indeed from a loop modification in the lid domain. The major changes in the functioning of MHETase from that of the other enzymes originate due to loop insertion, deletions, and recombination (Weber. G *et al.*, 2019). MHETase is coded by the gene ISF6\_0224. The gene was observed in the 104th contig out of the 227 contigs. The sequence that codes for the enzyme is as follows:

```
1381 ggcgccacgc tacgtggccc ctatcaggga ggcgccgccc agcgcaagt cgcttcggtg
1441 ttgatgtc cactaccctt gtagcgggag atttggggat acggcaaac tggcccgctg
1501 cgtgccca cccgaagta cctggcgtg cccgaccagg cggagatctg atccggcgc
1561 tcgccgctc cgaccagcc gaccaagggc gtcagcatg cgaaccgctc ggtcccggga
1621 ccgccgctc aatgattcat acccgccacc agaaaaagc gcgcgaatcc agcagacct
1681 ggcacgctc cgcaagtct ctcgtaatg tcggcctgt ccagggctga gaaccggcgc
1741 tcggacatg cgtgtacag gatacctt ccgccgctc cgcggaagc agcaaggtca
1801 gtagaagtg caccgtcca gtccatctg gattgggtga actgtccct ggtcggcaca
1861 atctcaacg gatgatgtc gaaatcgaat tcatcatcc gcgcccgaac ctgctgatg
```

1921 ggcatcggtt cggcggtgt ggcgaatcg accagccagg agcgcgccga gaagccgctc  
 1981 acgcgtggg cgttgtggc actgctgtg aagctaccga gccaccaga gccaccacc  
 2041 tggttatagg tctgccaact cagtcgctc atgccagcgt cccaagccca tctgtttgac  
 2101 aacggcgttc ccgcgaatt cacgggcccc gccatcgccg ctttgatcgc agtgacctgg  
 2161 acggggctca ggcaatccgc agtcttggcg ccaacgcaact ggagcgctg gccgtttgca  
 2221 ggattggctg ccgtggcggg gtcaagacct gcctggcagg cccggaat gtcaacgatg  
 2281 ccattccgaa gccctcagc ggcgtccat gtcccagga ttcccttga aagaaggtgc  
 2341 agtgcggcgt cagagaaaga ctattgata agcggcagc cttgtgcat caaacgaca  
 2401 cggcgggcgc caagggaact tgtgttccac gcaccgctga tcccggcctt gcgaactgg  
 2461 tagcctggtg caccggcaac gatgccctg tagtgggaag ggaagcctg cgagagcatc  
 2521 atgccctgc gcccgccctc gaaacagcgc atgaaatagg acttgcggc agcgcgtccg  
 2581 tagaacctgg ccaccgggc cttgctgccc tgcgtgacct gatcgtaca gttgtagccc  
 2641 atacggagac gggcctcgg atccaatccg aaggcaacgg tgccgagcgc gtcgggattg  
 2701 tcttctactg cgttctctg acctccatcc gttgcatgg tggcgaat gcgagacagc  
 2761 gcgcttgcga tctgtcccc accaatcctg ccggtagccg cggacagcga gccgttagtt  
 2821 ccgctgccg cttccatgaa gaaacgacc tccactggc cgggcatcc gagcggaat  
 2881 ttgattcgt agggatagcc gtcgatccg gtcgcttcg cgatggcgc ggacacctg  
 2941 cagtgttcgg gcagcgcgc gcgagatcg gtcggggcg cggcgtcag ccaggcagcg  
 3001 actccacga ccgtcggcg atttggccac accatgtcg ctttcctc cttcagacc  
 3061 tgcgatcgg cgcgcaagc caagggcacc gggggcggcg gaggtcctg ctgcggcgcc  
 3121 tctgtcggc gcagcggcgc atccaatccg aaggcaacgg tgccgagcgc gtcgggattg  
 3181 gacgcagca gcatcgtgtg gactgttgc tgcatttat gtcctctc actggttgc  
 The gene sequence is made up of 1811 base pairs. The translation of the above-mentioned nucleotide yields a result as seen below:  
 MQTTVTMLLASVALAACAGGGSTPLPLPQQQPEPPPPV  
 LASRAACEALKDGNMVPNAATVVEVAAWRDAAPATASAAAL  
 PEHCEVSGAIKRT  
 GIDGYPIEIKFRLRMPAEWNGRFFMEGGSGTNGLSAATGSIGGGQIA  
 SALSRNFATI  
 ATDGGHDNAVNDNPDALGTVAFLDLPQARLDMGYNSYDQVTQAG  
 KAAVARFYGRAADK  
 SYFIGSSEGGREGMMLLSQRFPSHYDGIVAGAPGYQLPKAGISGAWTT  
 QSLAPAAVGLD  
 AQGVPLINKSFSADLHLLSQAILGTCDALDGLADGIVDNYRACQAA  
 FDPATAANPAN  
 GQALQCVGAKTADCLSPVQVTAIKRAMAGPVNSAGTPLYNRWAWD  
 AGMSGLSGTTYNQ  
 GWRSWWLGSFNSSANNAQRVSGFSARSWLVD FATPPEPMPMTQVA  
 ARMMKFDFDIDPL  
 KIWATSGQFTQSSMDWHGATSTDLAAFRDRGGKMILYHGMSDAAFS  
 ALDTADYYERLG  
 AAMPGAAGFARLFLVPGMNHCSGGPGTDRFDMLTPLVAWVERGEA  
 PDQISAWSGTPGY  
 FGVAARTRPLCPYPQIARYKSGSDINTEANFACAAPP

There are a total of 600 amino acids in the sequence. The above-mentioned nucleotide and amino acid sequence can be obtained from NCBI using the accession ID: BBYR01000104.

The structure has been obtained from Uniprot, Accession ID: A0A0K8P8E7

The enzyme, as mentioned previously is a Hydrolase, more specifically chlorogenate esterase.

### 2.3 Cloning of PETase and MHETase

The functional expression of PETase can be carried out in green microalgae (*Chlamydomonas reinhardtii*). This alga is a unicellular photosynthetic microorganism. The ability to incorporate the production of the enzyme PETase is achieved through the transformation with the aid of a recombinant vector pBR9\_PETase\_Cre (Ji Won Kim *et al.*, 2020)<sup>[8]</sup>.

The functional expression of the MHETase enzyme can be expressed in E.coli BL21 with the aid of the vector pUCIDT plasmid, which also constitutes a native signal peptide derived from *Ideonella sakaiensis* and a constitutive promoter J23106 (Janatunaim, Rifqi Z. *et al.* 2020)<sup>[7]</sup>.

The two enzymes can then be combined to form a super enzyme complex. This work was carried out at the University of Portsmouth, U.K, under the guidance of Prof. John.E.Mc Geehan. The combining of the two enzymes generated a super enzyme complex that was capable of degrading the PET six times faster (Carrington, 2020)<sup>[2]</sup>.

### 2.4 Separation of Ethylene Glycol and Terephthalic acid

The action of the enzymes produced by the bacteria *Ideonella sakaiensis*, acts on and degrades Polyethylene terephthalate into MHET and then into two by-products, namely, Ethylene glycol and Terephthalic acid. Now, these two substances must be separated as they might be produced as a mixture.

An economical and rather environment-friendly method of separation of these two compounds was achieved by Gerald C. Tustin and his colleagues. The technique they developed involved the separation of terephthalic acid and ethylene glycol from a complex resin through a six-step method. Of this technique, only five steps is what we require. Cooling the mixture of these two compounds will leave the terephthalic acid in the solid phase and the ethylene glycol in the liquid phase. For obtaining the factory-grade forms of these compounds they have to be further treated. Ethylene glycol can be separated from the liquid portion using distillation techniques. The terephthalic acid can be recovered from the solid portion by heating it in the presence of water vapours at high enough temperatures. This acid-water vapour mixture can be cooled below the dew point of terephthalic acid. From this the polymer grade terephthalic acid can be isolated. (Tustin *et al.*, 1995)<sup>[17]</sup>.

### 2.5 Dealing with Ethylene Glycol

The ethylene glycol, which can be separated from the terephthalic acid, cannot be uptaken by many microbial species, owing to its high toxicity. For this reason, ethylene glycol must be converted into a less toxic form. To do this, it is treated with periodic acid which converts it into formaldehyde, which is arguably a less toxic form of ethylene glycol.

The reaction mechanism is named Glycol cleavage. It is a specific type of reaction in organic chemistry. This is a type of oxidation reaction.

As stated above, the ethylene glycol upon reacting with periodic acid can yield Formaldehyde as the reaction product. The formaldehyde so formed can be utilized by a variety of methylotrophic bacteria. This bacteria does not essentially utilize formaldehyde as a source of carbon. Instead, they utilize formaldehyde as a cofactor, for the degradation and assimilation of single carbon atom molecules, such as methane, methylamine, and dichloromethane, as a source of energy (Vorholt, 2002)<sup>[18]</sup>.

### 2.6 Dealing with Terephthalic acid

Terephthalic acid is an important industrial chemical. But if left untreated or exposed to water, it can act as a major pollutant. Recent studies have shown that terephthalic acid can be degraded using physical and chemical methods. But these techniques are rather expensive and complex. Because of these reasons microbial degradation of terephthalic acid is being considered as a viable alternative for the degradation of terephthalic acid. This is also an environmentally friendly method. This was achieved by utilizing a bacterial species known as *Arthrobacter sp.* Rather than utilizing the enzyme produced by the organism, the resting cells are being used for the purpose. This is because the extraction of the enzyme is a rather expensive method and also within the cells the enzymes will be protected from the external environment (Yi-Mei Zhang *et al.*, 2013)<sup>[19]</sup>.

### 3. Methodology

The study with an extensive literature search using Google Scholar and Google search engine using the search string “Microbial degradation Polyethylene terephthalate”.

#### 3.1 Cloning and expression of PETase enzyme in *Chlamydomonas reinhardtii*

1. The amino acid sequence of PETase (ISF6\_4831) obtained from UniProt.
2. The amino acid sequence of PETase can be reverse-translated, codon-optimized for *C. reinhardtii*, and synthesized by using the resulting pIDT\_PETase\_Opt. The codon-optimized PETase-encoding gene can be digested and cloned into pBR9\_mCherry\_Cre.
3. The digestion and cloning can be done by using XhoI and BamHI restriction endonucleases, generating pBR9\_PETase\_Cre.
4. With the transformation of PETase gene, expression confirmation, and plastic degradation, microalgal strains can be used.
5. For gene cloning, the ligated plasmid (PETase gene from pIDT\_PETase\_Opt + pBR9\_mCherry\_Cre) can be transformed into *E. coli* DH5 $\alpha$  by using commercial heat shock competent cells.
6. The heat-shocked cells can then be spread on LB-agar plates (BD) containing 100 mg/L Ampicillin and the correct clones can then be confirmed by polymerase chain reaction (PCR) and DNA sequencing. From the final *E. coli* clone, pBR9\_PETase\_Cre plasmid can be purified from the 300 mL culture solution by using Nucleobond Xtra Midi Plus for the transformation of microalgae.
7. Competent cells of *C. reinhardtii* CC-124 and CC-503 can then be prepared by using MAX<sup>TM</sup> Efficiency Transformation Reagent for Algae. Each 400  $\mu$ L of competent cells containing 10  $\mu$ g of pBR9\_PETase\_Cre (linearized by PsiI) is to be subjected to electroporation.
8. The electroporation can be performed with Gene Pulser Xcell Electroporation systems under the conditions of 500 V, 50  $\mu$ F, and 800  $\Omega$ .
9. After the pulse, 10 mL of TAP medium supplemented with 40 mM of sucrose is added to the cells immediately, and the cells are recovered for 16 h at 25 °C with shaking (80 rpm) under low light.
10. The cells can be plated on TAP-agar plates containing 10 mg/L Zeocin. Colonies from the TAP-agar plates can be inoculated into 96-well plates with 150  $\mu$ L TAP medium containing 10 mg/L Zeocin. After cultivation, cells in green-colored wells can be transferred to 24-well and 12-well plates sequentially using the aforementioned conditions. For the selection of stable transformants, the cells from 12-well plates can be cultured with a 10 mL TAP medium in a T-flask. The well-grown cells can be chosen as the final clones. To verify the PETase gene integration into the nuclear genome of *C. reinhardtii*, the genomic DNA of the final clones can be extracted via boiling with specific solution (1 M KCl, 10 mM EDTA, 100 mM of Tris-HCl, pH 9.5) for subsequent PCR analysis.
11. The protein expression can be analyzed by Western blot analysis and SDS-PAGE analysis.
12. To analyze the catalytic activity of PETase produced from *C. reinhardtii*, a commercial PET beverage bottle (PepsiCo., NY, USA) can be used as the substrate. The PET powder for HPLC analysis can be prepared by

grinding the bottle with sandpaper.

13. The ground PET was filtered using a sieve to remove large fragments. The PET films for electron microscopy were prepared by cutting a PET bottle into 2 cm  $\times$  1 cm fragments.
14. Wild-type *C. reinhardtii* were cultivated and harvested to obtain cell lysates. The harvested cells were resuspended in 1  $\times$  PBS and Protease Inhibitor Cocktail.
15. The solution was disrupted by sonication on ice for 20 min with 3 s pulses at 7 s intervals.
16. After the preparation of all materials, PET powder and films were incubated with 1 mL and 5 mL of the cell lysates, respectively.
17. All of the reaction mixtures were incubated at 30 °C for 4 weeks, changing the cell lysate to a fresh one weekly to prevent contamination and PETase degradation (Ji Won Kim *et al.*, 2020) [8].

#### 3.2 Cloning and expression of MHETase in *E. coli* BL21

**Methods:** Construction of MHETase gene in pUCIDT plasmid with native signal peptide from *I. sakaiensis* 201-F6 and constitutive promoter J23106 can be expressed in *Escherichia coli* BL21 (DE3) by heat shock. Expression analysis using SDS-PAGE and activity of enzymes is analyzed by spectrophotometry method and SEM.

**Results:** MHETase gene protein can be successfully constructed in pUCIDT +Amp plasmid with native signal peptide from *Ideonella sakaiensis* 201-F6, T7 terminator, and constitutive promoter J23106. PCR analysis showed that the gene was successfully contained in the cells by band size (1813 bp) in electrophoresis gel. Analysis using SnapGene, pairwise alignment using MEGA X, and NCBI can be demonstrated that in the MHETase sequence the gene can be in-frame in pUCIDT plasmid.

**Conclusion:** MHETase gene can be successfully constructed in plasmids by the in-silico method. Synthetic plasmids transformed in *E. coli* BL21 contain MHETase gene sequences which were in the frame. Hence, the *E. coli* BL21 cells have the potential to produce MHETase proteins for the plastic degradation testing process (Janatunaim, Rifqi Z., *et al.* 2020) [7].

#### 3.3 Separation of the products of enzymatic degradation by bacterial enzymes

- (1) Cooling the mixture with water at about 70° C. to 100° C., filtering the solids, washing the solids, and then drying the solids at a temperature of from about 25 C. to 100° C. to provide a solid portion comprised of the terephthalic acid and a liquid portion composed of ethylene glycol.
- (2) Recovering the ethylene glycol from the liquid portion of the mixture by a two-step distillation, wherein in a first step of the two-step distillation the water and low boiling components are removed at about 0.1 to 6 atmospheres pressure and temperatures of about 100 °C. to 170 °C., and wherein in a second step of the two-step distillation high boiling species are removed at about 1 mmHg to 10 atmospheres pressure and at a temperature range of about 50 °C. to 300 °C.
- (3) Recovering the solid terephthalic acid by heating the solid portion above its dew point with a continuous stream of water vapor at a temperature of about 310 C. to 370° C. and a pressure of about 0.1 atmospheres to 1.2 atmospheres to produce a vapor comprised of the water and terephthalic acid.
- (4) Cooling the vapor containing the terephthalic acid and the water to a temperature below the dew point of 2 the terephthalic acid and.

(5) Collecting the solid polymer grade terephthalic acid (Tustin *et al.*, 1995) <sup>[17]</sup>.

#### 4. Expected results and discussion

##### 4.1 Cloning and expression of PETase enzyme in *Chlamydomonas reinhardtii*

Firstly, expect to find the stable integration of the PETase encoding gene in the plasmid, pBR9\_mCherry\_Cre, and subsequently leading to the production of the stable recombinant plasmid pBR9\_PETase\_Cre. After this, it is expected that this plasmid can be successfully transferred to an *E. coli* DH5 $\alpha$  host. Following this, the *E. coli* should be able to infect the *Chlamydomonas reinhardtii* and transfer the gene of interest into the nuclear genome of the *Chlamydomonas reinhardtii*. The *Chlamydomonas* should then be able to produce the enzyme PETase in an economically viable amount.

##### 4.2 Cloning and expression of MHETase enzyme in *E. coli* B12

Here in this stage, expect to see the stable integration of the MHETase coding gene into the plasmid pUCIDT +Amp. From there expected that this recombinant synthetic plasmid can then be stably transferred into an *E. coli* B12 expression vector. Suppose to see that *E. coli* can produce the enzyme MHETase in an economically viable amount.

##### 4.3 Separation products of enzyme degradation

Following the enzymatic degradation of PET, left with Terephthalic acid and Ethylene glycol in the mixture. Following a variety of procedures, expected to separate the two components from the mixture with the maximum purity as much as it is possible.

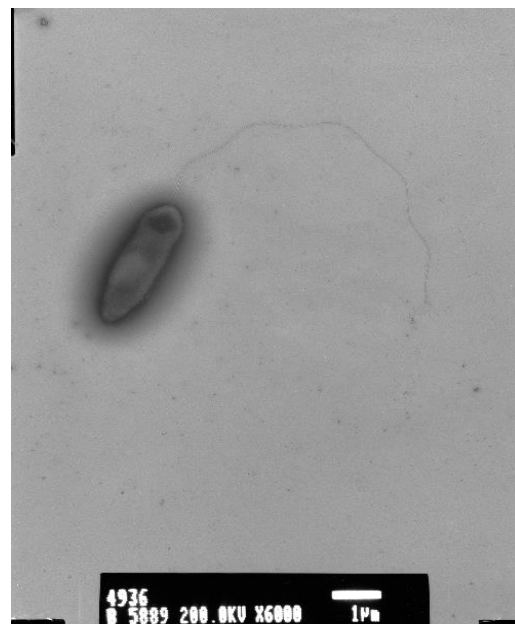
#### 5. Conclusions

Plastic pollution is a menace that is eating away at our planet with each passing day. With this review literature, we aim to bring to light a possible method to achieve the complete degradation of PET plastics, without involving any of the physical damage that any of the conventional methods would bring forward. Microbial degradation is the only solution to this problem. With the aid of the microbe *Ideonella sakaiensis* and the enzymes that it is capable of producing, the complex

substance PET can be converted into much simpler forms, which can then be utilized by other organisms as a source of energy or potentially use it as a catalyst.

Some of the few limitations that can be faced while utilizing this method there can be some limitations. They are as follows:

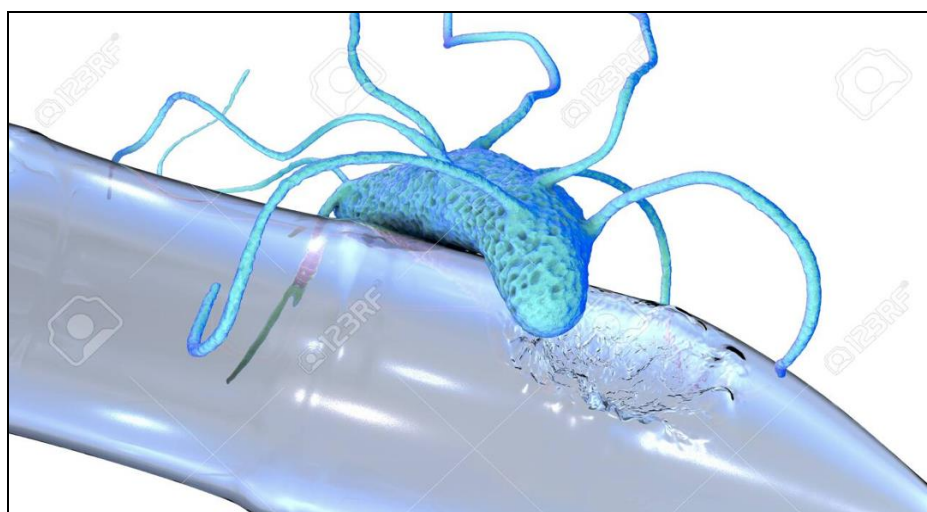
1. *Ideonella sakaiensis* is native to Japan and it is rather costly to obtain, needing very specific growth medium components. This could pose a challenge.
2. The exact economic feasibility of the entire work is not determined.
3. The duration it takes to attain the complete biodegradation of PET is not exactly known. Depending on the size of the plastic pollutant the duration of the process might vary.



Source:

<https://www.microbiologyresearch.org/content/journal/ijsem/10.1099/ijsem.0.001058#tab2>

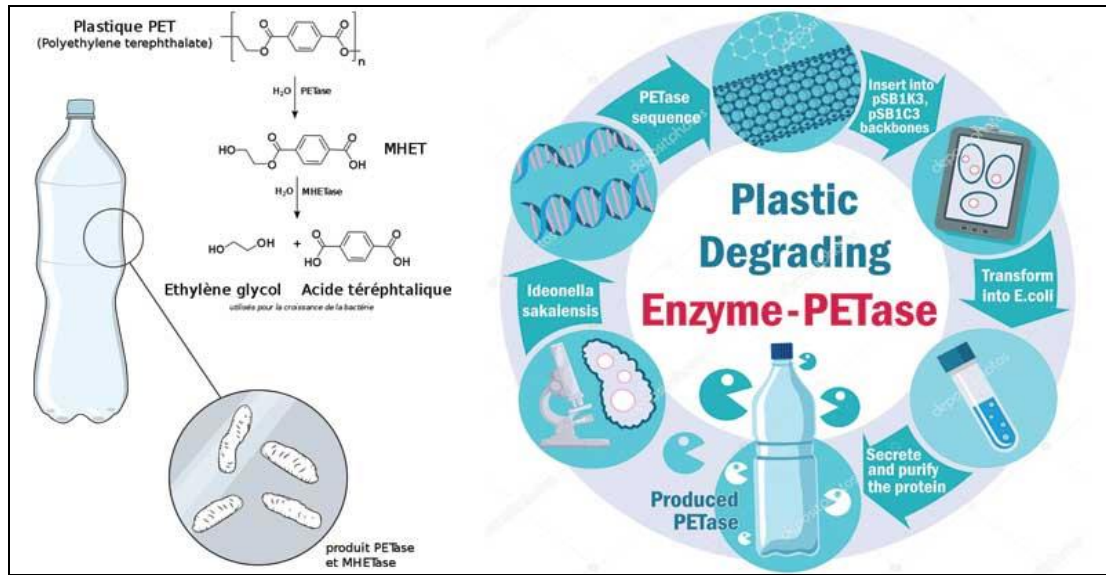
**Fig 1:** Transmission electron micrograph of a cell of strain 201-F6T grew in NBRC no. 802 broth at 30 °C for 24 h. Bar, 1  $\mu$ m.



Source: [https://www.123rf.com/photo\\_127986519\\_stock-illustration-plastic-degrading-bacteria-ideonella-sakaiensis-3d-illustration-recently-discovered-bacteria-that-ha.html](https://www.123rf.com/photo_127986519_stock-illustration-plastic-degrading-bacteria-ideonella-sakaiensis-3d-illustration-recently-discovered-bacteria-that-ha.html)

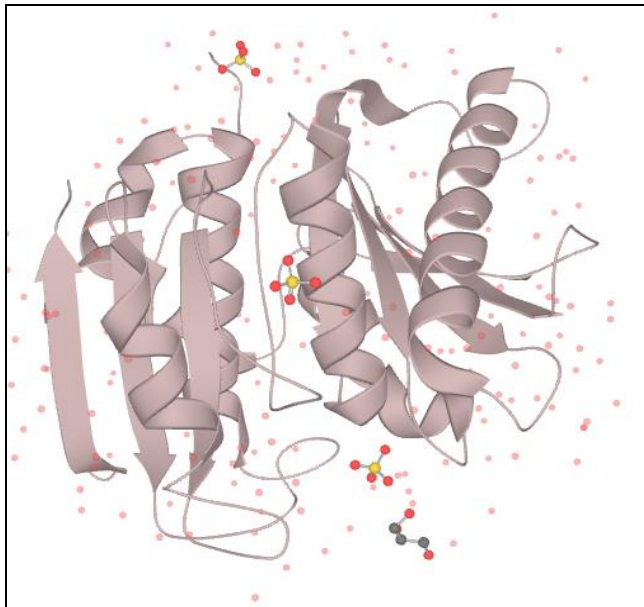
**Fig 2:** *Ideonella sakaiensis*, 3D illustration





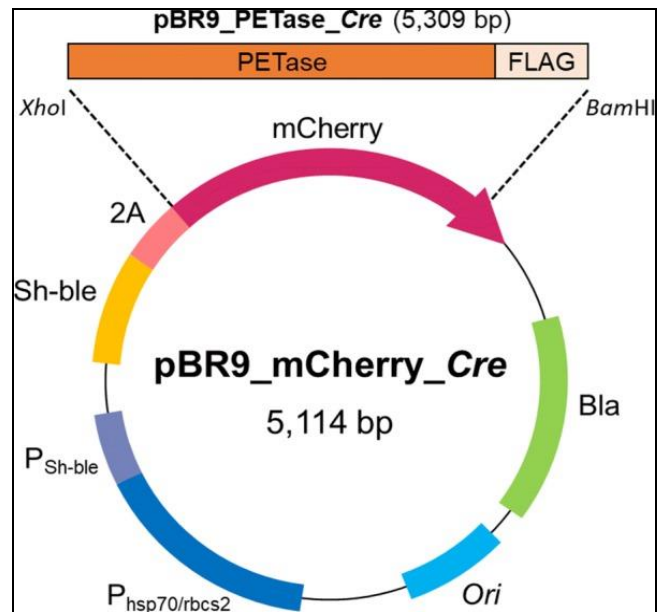
Source: <https://thebiologynotes.com/ideonella-sakaiensis-plastic-eating-bacteria/>

Fig 3: Action of *Ideonella sakaiensis* on plastics



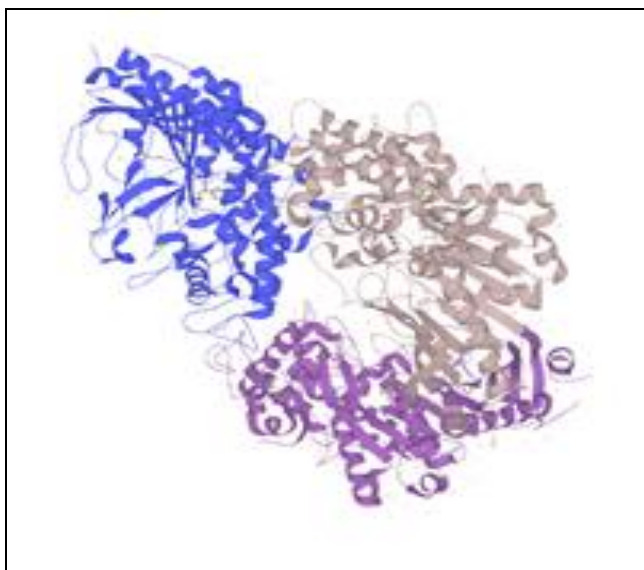
Source: <https://www.uniprot.org/uniprot/A0A0K8P6T7>

Fig 4: Tertiary structure of the enzyme PETase.



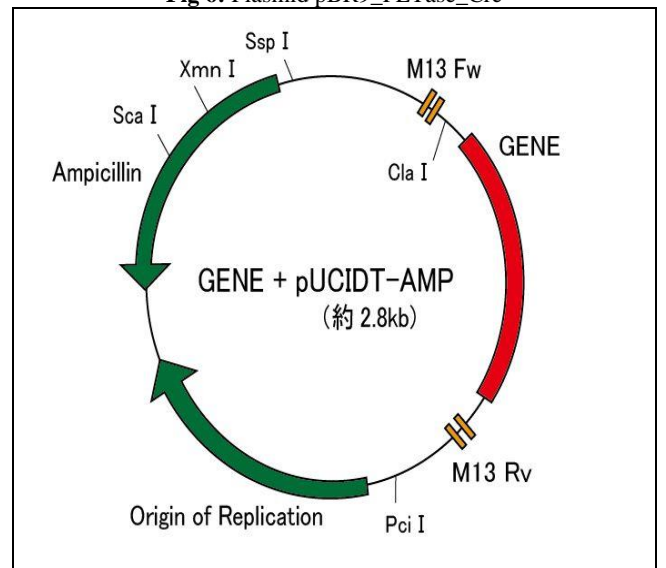
Source: <https://www.ncbi.nlm.nih.gov/pmc/articles/PMC7189453/>

Fig 6: Plasmid pBR9\_PETase\_Cre



Source: <https://www.uniprot.org/uniprot/A0A0K8P8E7>

Fig 5: Tertiary structure of the enzyme MHEase.



Source: [https://sg.idtdna.com/jp/site/custom\\_gene.html](https://sg.idtdna.com/jp/site/custom_gene.html)

Fig 7: Recombinant pUCIDT plasmid containing gene of interest.

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**Conflicts of interest:** The author do not have any conflict of interest.

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